

Designation: D7906 - 22

Standard Practice for Recovery of Asphalt Binder from Solution Using Toluene and the Rotary Evaporator¹

This standard is issued under the fixed designation D7906; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This practice is intended to be used to recover asphalt from a solvent using the rotary evaporator to ensure that changes in the asphalt properties during the recovery process are minimized.

1.2 *Units*—The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.3 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.

1.4 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.

2. Referenced Documents

2.1 ASTM Standards:²

- D92 Test Method for Flash and Fire Points by Cleveland Open Cup Tester
- D2172/D2172M Test Methods for Quantitative Extraction of Asphalt Binder from Asphalt Mixtures
- D3666 Specification for Minimum Requirements for Agencies Testing and Inspecting Road and Paving Materials
- E1137/E1137M Specification for Industrial Platinum Resistance Thermometers

3. Summary of Practice

3.1 The solution of solvent and asphalt from a prior extraction is first distilled by partially immersing the rotating distillation flask of the rotary evaporator in a heated oil bath while the solution is subjected to a partial vacuum. During the final stage of the process, vacuum is increased and nitrogen or carbon dioxide gas is introduced to the solution to remove remaining toluene and prevent further aging of the asphalt. The recovered asphalt can then be tested as required.

4. Significance and Use

4.1 To determine the characteristics of the asphalt in an asphalt paving mixture, it is necessary to extract the asphalt from the aggregate by means of a suitable solvent and then to recover the asphalt from the solvent without significantly changing the asphalt's properties. The asphalt recovered from the solvent by this practice can be tested using the same methods as for the original asphalt cement, and comparisons between the properties of the original and recovered asphalt can be made.

4.2 Users are cautioned that the recovered binder properties from an asphalt mixture sample may not exactly represent the properties of the original asphalt binder due to factors outside of laboratory control such as aging, contamination, and molecular changes caused by exposure to heat and solvents. Consequently, recovered asphalt binder properties should not be used as a basis for acceptance.

Note 1—The quality of the results produced by this is dependent on the competence of the personnel performing the procedure and the capability, calibration, and maintenance of the equipment used. Agencies that meet the criteria of Specification D3666 are generally considered capable of competent and objective testing, sampling, inspection, and so forth. Users of this practice are cautioned that compliance with Specification D3666 alone does not completely ensure reliable results. Reliable results depend on many factors; following the suggestions of Specification D3666 or some similar acceptable guideline provides a means of evaluating and controlling some of those factors.

5. Apparatus

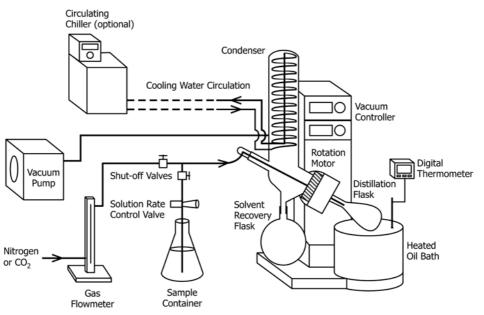
5.1 *Rotary Evaporator* (see Fig. 1), equipped with distillation flask, variable speed motor capable of rotating the distillation flask at a rate of at least 40 rpm, condenser (Note 2), solvent recovery flask, and heated oil bath. The angle of the

¹ This practice is under the jurisdiction of ASTM Committee D04 on Road and Paving Materials and is the direct responsibility of Subcommittee D04.25 on Analysis of Asphalt Mixtures.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

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NOTE 1—The vacuum pump and vacuum controller are recommended for maintaining the required vacuum in 5.1. The diagram displays a Büchi model rotary evaporator.

FIG. 1 Rotary Evaporator and Recovery System

distillation flask from the horizontal to the bath is set at approximately $30 \pm 10^{\circ}$. The distillation flask when fully immersed in the oil bath should be at a depth of approximately 40 mm.

5.2 Distillation and Receiving Flasks, a distillation flask with a minimum capacity of 1 L is required for the sufficient recovery of asphalt binder. It is recommended for the receiving flask to have a minimum capacity of 1 L to maximize the amount of solvent recovery.

Note 2—A vertical condenser is recommended for solvents with higher boiling points such as toluene.

5.3 *Vacuum Pump and Controller*, capable of maintaining a vacuum of 72.0 ± 0.7 kPa down to 6.7 ± 0.7 kPa.

5.3.1 *Vacuum Controller*, built into rotary evaporator system controlling and measuring the specified vacuum.

Note 3—Alternatively, a house vacuum system and vacuum gauge may be used if the rotary evaporator system does not include vacuum pump and controller.

5.4 *Gas Flow Meter*, capable of indicating a gas flow of up to 1 L/min.

5.5 *Sample Container*, having adequate volume to hold the sample and added solvent.

5.6 *Thermometer*, capable of measuring the oil bath temperature up to 180 °C with an accuracy of ± 1.5 °C (2.5 °F) at a depth of 40 \pm 5 mm. The thermometer shall be one of the following:

5.6.1 A digital metal stem thermometer with a thermocouple sensor and a stem length of at least 150 mm (6 in.) paired with an appropriate meter capable of displaying temperature to the nearest 0.1 °C (0.2 °F). The sensor shall be encased in a stainless steel sheath that has a length of at least 150 mm (6 in.) and a minimum immersion depth of not more than 40 mm (1.6 in.).

5.6.2 A Class A Pt-100 RTD (Specification E1137/E1137M) sensor with a three or four-wire configuration at the connection terminal and paired with an appropriate meter capable of displaying the temperature to the nearest 0.1 °C (0.2 °F). The sensor shall be encased in a stainless steel sheath that has a length of at least 150 mm (6 in.) and a minimum immersion depth of not more than 40 mm (1.6 in.).

5.7 *Circulating Chiller*, capable of cooling circulating water to increase condensation of toluene.

Note 4—Circulating chiller is not required if condensing water is normally below 25 °C, as per manufacturer's recommendation. Colder circulating water temperatures below 25 °C can increase the rate of condensation of the toluene vapor.

6. Reagents and Materials

6.1 *Purity of Reagents*—Reagent-grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.³ Other grades may be used provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

6.2 *Nitrogen Gas or Carbon Dioxide Gas*—A pressurized tank with pressure-reducing valve or other convenient source.

6.3 *Oil*—The oil for the heated oil bath should be USP White Oil, Silicone Fluid SWS-101 with flash point above

³ Reagent Chemicals, American Chemical Society Specifications, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.